

ALKALOIDAL COLOR REACTIONS

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Alkaloidal Color Reactions

One of the properties by which alkaloids are characterized is that of producing colored compounds when treated with various reagents. This is a property by no means peculiar to alkaloids, but it is sufficiently characteristic to frequently serve as a means of detecting the presence of alkaloids; and many of the reactions are so specific that they serve for the identification of individual alkaloids. Indeed, the final identification of most alkaloids is by means of their color reactions.

In most cases little appears to be known of the products formed by means of these color producing reagents other than that they are colored. However, most of the reagents employed to bring about the color reactions may be placed in two classes: oxidation reagents and dehydration reagents. Among the former may be included nitric acid, iodic acid, potassium dichromate, potassium permanganate, halogens, formaldehyde, etc. Among the latter, sulphuric acid, zinc chloride and acid anhydrides. Other reactions which may produce a distinct change in color are those which involve the formation of salts with considerable change in the molecular formula. We know that some alkaloids which are colorless in themselves form colored salts from which the colorless alkaloid may be regenerated. Upon the whole, however, little is known of the nature of the colored product.

Table I

ALKALOIDS	REAGENTS		
	Perru- theniate	Selenous Acid in conc. H ₂ SO ₄	Sugar Titanic Anhydrides
Aconitine		yellowish	orange
Adrenaline			reddish-brown
Apomorphine		violet	reddish-violet
Aspidospermine			blue-violet
Atropine		colorless	
Atropine sulphate			
Berberine			green
Brucine		yellow-red	
Caffeine		colorless	
Cantharidin			
Chelidonine	green		
Cinchonamine			yellow
Cinchonidine			
Cinchonine			
Cocaine		colorless	
Codeine		blue-green	deep-pink rose or red
Colchicine		lemon yellow	
Coniine		colorless	
Cubebin			
Cupreine			
Curarine			
Delphinine		red brown	rose or red
Digitalin		yellowred	rose or red
Digitoxin			green

Table I cont'd

ALKALOIDS	Perru- theniate	REAGENTS Selenous Acid in conc. H ₂ SO ₄	Sugar	Titanic Anhydride
Emetine				rose or red
Ergotinine				green
Esserine				yellow
Heroin				blue-violet
Hordenine				deep orange
Hydrastine				rose or red
Imperatorin	blue-green			
Lobeline				blue-violet
Morphine		blue-green	light pink	blue to blood red
Morphine HCl				
Narceine		greenish yellow		yellow
Narcotine		greenish-blue		
Nicotine		yellowish		rose or red
Onanin	brownish red			
Oxydimorphine				wine red
Papaverine		greenish blue		blue-violet
Physostigmine		brownish yellow		
Picrotoxin		colorless		yellow
Pilocarpine				
Piperine				
Podophyllotoxin				yellow
Quinidine				
Quinine		colorless		yellow

Table I cont'd

ALKALOIDS	REAGENTS			
	Perru- theniate	Selenous Acid in conc. H ₂ SO ₄	Sugar	Titanic Anhydride
Quinine sulfate				
Salicin				
Santonin				
Saponin				
Sparteine				
Senegin				
Solanine	red	reddish yellow		
Strychnine		colorless		yellow
Thebaine		deep orange		
Tyrosine				
Veratrine		lemon yellow- olive green	dark red- blue	rose or red

Table II

ALKALOIDS	REAGENTS			
	Zinc Chloride	Furfural with H_2SO_4	Chloral hydrate with H_2SO_4	Ammonium Sulpho-selenite
Aconitine				rose
Adrenaline				
Apomorphine				
Aspidospermine				
Atropine				
Atropine sulphate				
Berberine	yellow			yellow-red reddish
Brucine				
Caffeine				
Cantharidin				
Chelidonine				
Cinchonamine				
Cinchonidine				
Cinchonine				
Cocaine				
Codeine		red brown	green	
Colchicine				
Coniine				
Cubebine	crimson			
Cupreine				
Curarine				violet

Table II cont'd

ALKALOIDS	REAGENTS			Ammonium sulpho-selenite
	Zinc chloride	Furfural with H_2SO_4	Chloral hydrate with H_2SO_4	
Delphinine	brown red			reddish
Digitalin	chestnut brown	brown		yellowish
Digitoxin				
Emetine				
Ergotinine				
Esserine				lemon yellow
Heroine				
Hordenine				
Hydrastine				
Imperatorin				
Lobeline				
Morphine <small>HCL</small>		red brown	green	greenish blue
Narceine	olive green		red	yellowish green
Narcotine			yellowish green	bluish
Onanin				
Oxydimorphine				
Papaverine		brownish	violet	bluish green
Physostigmine				
Picrotoxin			red	
Pilocarpine				
Piperine				
Podophyllotoxin				

Table II cont'd

ALKALOIDS	REAGENTS			Ammonium sulpho-selenite
	Zinc chloride	Furfural with H_2SO_4	Chloral hydrate with H_2SO_4	
Quinidine				
Quinine	pale green	brown		
Quinine sulfate		green		
Salicin	reddish-violet			
Santonin	bluish violet			
Saponin				yellowish
Sparteine				
Senegin				dirty yellow
Solanine			red brown	canary yellow
Strychnine	bright red	rose	dirty brown	yellow
Thebaine	yellow			red
Tyrosine				
Veratrine	red	yellow-greenish blue	red	

Table III

ALKALOIDS	REAGENTS			
	Ammonium uranate	Chlorinated lime	p-dimethyl amido-benz- aldehyde	Sulfuric Acid
Aconitine		greenish yellow		yellowish brown
Adrenaline				
Apomorphine				
Aspidospermine				
Atropine				
Atropine sulphate				
Berberine				
Brucine		light green		pink
Caffeine				green
Cantharidin				
Chelidonine				
Cinchonamine				
Cinchonidine		yellow		
Cinchonine		yellow		
Cocaine				
Codeine	blue		red	
Colchicine				
Coniine				
Cubebin				
Cupreine				
Curarine				

Table III cont'd

ALKALOIDS	REAGENTS			Sulfuric acid
	Ammonium uranate	Chlorinated lime	p-dimethyl amido-benzaldehyde	
Delphinine				
Digitalin				black brown
Digitoxin				
Emetine				brown
Ergotinine				
Esserine				
Heroine				
Hordenine				
Hydraatine				
Imperatorin	blue			
Lobeline				
Morphine	dirty green		red	light pink
Morphine HCl				
Narceine				
Narcotine			orange	
Nicotine				
Onanin				
Oxydimorphine				
Papaverine			orange	
Physostigmine			green	
Picrotoxin				
Pilocarpine				
Piperine				blood red

Table III cont'd

ALKALOIDS	REAGENTS			
	Ammonium uranate	Chlorinated lime	p-dimethyl amido-benz- aldehyde	Sulfuric acid
Podophyllotoxin				
Quinidine				
Quinine			red brown	
Quinine sulfate				
Salicin				blood red
Santonin				
Saponin				
Sparteine				
Senegin				
Solanine				
Strychnine				
Thebaine				
Tyrosine				
Veratrine		light green	green	deep red

Table IV

ALKALOIDS	REAGENTS			
	Sulfuric acid and sugar	Potassium dichromate	Paraldehyde with H_2SO_4	Euchlorine ($KClO_3$ and HCl)
Aconitine		green		
Adrenaline				
Apomorphine			violet	
Aspidospermine				
Atropine		dark green		
Atropine sulphate	violet			
Berberine				
Brucine		bright red		violet
Caffeine		dark green		
Cantharidin				
Chelidonine				
Cinchonamine				
Cinchonidine		green		
Cinchonine		green		
Cocaine				
Codeine	cherry red	black	orange	
Colchicine				
Coniine				
Cubebin				
Cupreine				
Curarine				
Delphinine				

Table IV cont'd

ALKALOIDS	REAGENTS			Euchlorine (KClO ₃ and HCl)
	Sulfuric acid and sugar	Potassium dichromate	Paraldehyde with H ₂ SO ₄	
Digitalin		green		
Digitoxin				
Emetine		green		
Ergotinine				
Esserine				
Heroine				
Hordenine				
Hydrastine				
Imperatorin				
Lobeline				
Morphine		dirty brown	orange	
Morphine HCl	rose			
Narceine				
Narcotine	mahogany			yellow
Nicotine				
Onanin				
Oxydimorphine				
Papaverine			yellowish red	
Physostigmine				
Picrotoxin			yellow	
Pilocarpine				
Piperine		very dark		
Podophyllotoxin				

Table IV cont'd

ALKALOIDS	REAGENTS			
	Sulfuric acid and sugar	Potassium dichromate	Paraldehyde with H_2SO_4	Euchlorine ($KClO_3$ and HCl)
Quinidine		green		
Quinine		green		yellowish pink
Quinine sulfate	greenish			
Salicin	bright red	dark		
Santonin				
Saponin				
Sparteine				
Senegin				
Solanine			orange	
Strychnine	reddish	deep violet		red
Thebaine				
Tyrosine				
Veratrine	dark green	reddish brown	red brown	

BIBLIOGRAPHY

- Jorissen, A. 1880
 Chem. Zeit. - (New Rem. April p.112, Proceedings A.Ph.A.28, p.321)
 Conclusion that certain colors which certain alkaloids produce when treated with concentrated acids depended on withdrawal of water, led to the experimenting with pure (chemically) $ZnCl_2$, which is extremely hygroscopic. (Alkaloids must be present singly).
- Hamlin, Ben B. 1881
 Am. Journal Pharm. June pp.283-85. (Proceedings A.Ph.A. 29, p.324)
 Effect of concentrated sulphuric acid and subsequent effect of oxidizing agents (minute quantity of bichromate of potassium) (Solution of chlorinated lime). H_2SO_4 used probably contained traces of HNO_3 , which appears to have modified color in several instances.
- Robin, Maurice. 1881
 Revue Scientifique - (Am. Jour. Pharm. June, pp.283-85; Proceedings A.Ph.A. 29, p.324)
 Observations of the effect of H_2SO_4 in presence of sugar.
- Signor Czumpelitz 1882
 (Am. Jour. of Pharmacy 54, p.84)
 $ZnCl_2$ used to distinguish alkaloids.
- (Am. Jour. Pharm. 60, 0.247) 1888
- Brociner, A.L. 1890
 Jour. de Pharm et de Chim. XX, p.390 (Am. Jr. Pharm. Feb. p.94, Proceedings A.Ph.A. 38, p.665)
 1 gm Perruthenate or ruthenate of Potassium in 20 cc. very pure concentrated sulphuric acid (Sp.Gr. 1.184)
- Ferreira de Silva, A.J. 1891
 Comptes Rendus, pp.112, 1266; (Jr. Anal & Applied Chem. Aug. 1891, p.478; Am. Jr. Pharm. 63, pp.502-3.)
 Ammonium sulphoselenite as reagent, (prepared by Lafon and used by author)
- Dr. Mecke 1900
 Ztschr. f. Off Chem. 1899, p.351 (Apoth. Ztg. Sept. 6, 1899 p.532; Proceedings A.Ph.A. 48, pp.808-9)
 Selenous acid (0.5 gm) in (100 gm) conc. H_2SO_4 .

- Ruddiman, Edsel, A. 1903
Drug Circ. May and June pp.95-97, 119-123 (Proceedings A.Ph.A.51, p.948)
- Schaer, Prof. E. 1911
Arch.D. Pharm. 248 (1910) no.6, p.458 (Proceedings A. Ph.A. 59, p.473)
Mixture of 1 vol. perhydrol (30% sol.H₂O₂) and 10 vol conc. H₂SO₄
- Peset Buendia 1916
Anales Soc. espan.fisica quimica; (Jr. Pharm.Chim, 14, p.86; Yr.Bk.A.Ph.A. 5, p.369)
- Deniges, G. 1916
Ann.Chim. Anal, 21, p.213; (Pharm.Jr. -; Jr. A.Ph.A. 6, p.42)
The use of titanitic anhydride for the identification of alkaloids containing phenolic groups, suggested by the use of thymol as a reagent for titanium.
- Pharmazev. J. -, (Drug.Civi. 61, p.189; Yearbook A.Ph. A. 6, pp.418-19) 1917
Alkaloidal reagent obtained by dissolving 2 gms of p-dimethyl amidobenzaldehyde in 6 gms concentrated sulphuric acid and adding to solution 0.4 gm of water (solution keeps for two weeks).

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